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Thermodynamics of alloys, phase transformations in alloys, solidification, non-equilibrium processing, metallic glasses, metastable compounds, quasicrystals, high temperature alloys, high entropy alloys, nanoporous metals by alloy corrosion. metal oxidation.



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University of Amsterdam Van der Waals-Zeeman Institute, Amsterdam, Netherlands

Solid State Physics; Magnetism; Physical Metallurgy;



Hongge Pan

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Hydrogen storage materials and hydrides, Anode and cathode materials for rechargeable batteries, Supercapacitor, Magnetic materials, Photocatalytic materials



Vitalij Pecharsky

Iowa State University Department of Materials Science and Engineering, Ames, Iowa, United States

Structure-property relationships; Intermetallic and rare earth compounds. Electronic, magnetic and caloric materials. Mechanochemistry.

Editors



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Microstructures of alloys, Phase transformations in alloys, Electron microscopy (TEM/STEM), X-ray/electron diffraction, Crystallography of alloys/inorganic compounds



Mehmet Acet

University of Duisburg-Essen Faculty of Physics, Duisburg, Germany

Phenomena involving the interplay between magnetism and structure: magnetovolume effects (invar, anti-invar), magnetostructural transitions (Heuslers, anti-perovskites, manganites, and crystallographic properties at interfaces separating different magnetic configurations (shell-ferromagnets); functionalities relevant to refrigeration, energy-conversion, non-volatile magnetic memory, permanent magnets.



Jennifer Aitken

Duquesne University Department of Chemistry and Biochemistry, 600 Forbes Avenue, 308 Mellon Hall, Pittsburgh, Pennsylvania, 15282, United States

Solid-state chemistry, flux synthesis, chalcogenides, nonlinear optical materials, thermoelectrics, single crystal X-ray diffraction, powder X-ray diffraction, semiconductors



Na Chen

Tsinghua University School of Materials Science and Engineering, 100084, Beijing, China

Metallic glasses; Bulk metallic glasses; Glass nanocomposites; Thermodynamics of alloys; Non-equilibrium processing; High entropy alloys; Magnetic thin films



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The University of Sydney, Sydney, 2006, New South Wales, Australia



Lawrence Cook

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High temperature materials, Mechanical properties, Phase equilibria, Thermal analysis



Daria Drozdenko

Charles University Faculty of Mathematics and Physics Department of Physics of Materials, 3 Ke Karlovu, 121 16, Praha, Czech Republic
Analysis of plastic deformation in metals by acoustic emission (AE) technique., Complex study of Mg alloys (including Mg-LPSO-based alloys),
microstructure and mechanical properties, Advanced techniques for microstructure analysis



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structure refinement, degassing, exfoliation, metal processing, solidification



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Functional Ceramics, Nano Materials, Thin Films

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Phase Diagrams; Measurement of Thermochemical Properties; Solid State Ionics; Chemical Sensors and Sensor Materials; Hydrogen in Metals;
Chemical Synthesis of Inorganic Compounds



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Low Dimensional Structures and Devices, Nanotechnology and Nanoscience, Self-Assembled Semiconductor Nanostructures, Semiconductor Materials,
III-V Electronic and Optoelectronic Devices, Photovoltaic Materials and Devices, Molecular Beam Epitaxy, Deep Level Transient Spectroscopy



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Hydrogen research, Hydrogen storage, Metal hydrides, Gas-solid interactions, Materials characterization, Neutron diffraction, Materials synthesis



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Texture of Mg alloys, Texture induced deformation behaviour, Metal forming, Application of light alloys



Yongchang Liu

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Solid-state phase transformations: thermodynamic and kinetic analyses; rapid solidification; metal-activated sintering and Microstructural control in ni(co)-based superalloys, heat-resistance steels, superconducting materials, lead-free solders and ti-al intermetallic compounds.



Nicoleta Lupu

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Metallic glasses; Bulk metallic glasses; Magnetic and magnetoelectric materials; Magnetoelastic processes; sensors and devices; Physics and chemistry of surfaces and interfaces; Nanoparticles and nanowire arrays; Hydrogen storage materials.



Valmor Mastelaro

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Structure-property relationships, ZnO Based Materials, Metal Oxide Gas sensors, Glass and Glass-Ceramics, Metal oxide thin films, XAS and XPS spectroscopies



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Physical metallurgy, alloy design, Phase transformations, High entropy alloys, bulk metallic glasses, nanocrystalline materials, metal matrix composites, electron microscopy, atom probe tomography.



Hari Srikanth

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Magnetism and magnetic materials, Nanostructured materials for energy and biomedical applications, Structure-property correlations in functional materials, Strongly correlated systems



Wieslaw Strek

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Rare earth ions and transition metal ions, doped sol-gel materials, photonic structures, nanomaterials, nanoceramics and crystals.



Isabel Van Driessche

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Chemical Solution deposition (CSD, ink jet printing) of ceramics. Materials of interest : superconducting perovskites and buffer layers for production of coated conductors, titanates for (photo)catalytic and battery applications, low-E coatings; Formulation of environmentally friendly based inks. Use of bottom-up chemical synthesis approaches (hydrothermal, microwave-assisted, hot injection) for the synthesis of ceramic nanoparticles/suspensions.



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Experimental Condensed Matter: Magnetism, Magnetic Materials, Spintronics, and Spin Caloritronics



Renbing Wu

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Semiconductor, Transition Metal-based composites, Electrode materials for energy storage and conversion



Xuezhong Xiao

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Volodymyr Yartys

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Nanomaterials for energy storage. Rechargeable Batteries. Hydrogen as an Energy Carrier. New Intermetallics and Carbon Materials for Hydrogen Storage and Battery Applications. Synchrotron and neutron powder diffraction. Crystal structures of novel materials.

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Self-propagating High-temperature Synthesis (SHS), Transition metal borides and nitrides, Intermetallics, MAX phases, Thermite Reaction

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Structural and multiferroic properties in double-layer Aurivillius phase $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ prepared by molten salt method

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ABSTRACT

A single-phase sample of the Aurivillius compound $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ was prepared by a molten salt method using $\text{K}_2\text{SO}_4/\text{Na}_2\text{SO}_4$ as the flux. The crystal structure, morphology, ferroelectric, and magnetic properties were investigated. Neutron powder diffraction data confirmed a non-centrosymmetric orthorhombic crystal structure with space group $A2_1am$ and Pb/Bi disorder in the bismuth oxide blocks, Bi/Pb/La disorder on the perovskite A-site, and Nb/Mn disorder on the perovskite B-site. The morphology of the sample showed anisotropic plate-like grains as probed by scanning electron microscopy. The dielectric constant exhibits a transition peak between 600 K and 640 K that depends on frequency, indicating relaxor ferroelectric behavior. Electrical polarization versus applied field loops are unsaturated, with a remnant polarization of $0.43 \mu\text{C}/\text{cm}^2$ at 40 Hz under the maximum electrical field applied of 160 kV/cm. The ferroelectricity originates from the displacement of oxygen atoms in the BO_6 octahedra, resulting in a polar structural distortion. Magnetic susceptibility measurements showed the presence of mixed Mn^{3+} and Mn^{4+} , resulting in short-range ferromagnetic order via double exchange interactions below 33 K. The remnant magnetization (M_r) is 0.01 emu/g at 5 K. This mixed valence of Mn cations is mainly responsible for the high electrical conductivity. Thus, $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ exhibits coexisting ferroelectric and ferromagnetic properties.

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1. Introduction

Multiferroic compounds presenting the coexistence of ferroelectric and ferromagnetic order have attracted a lot of attention for several decades due to their interesting properties and potential applications for data storage devices (RAM), sensors, spintronic and other electronic devices [1–3]. Among them, the Aurivillius compounds are of particular interest due to their structural flexibility. The Aurivillius structure is constructed from alternating perovskite-like blocks and bismuth oxide blocks and can be represented by the general formula $(\text{Bi}_2\text{O}_2)^{2+}(\text{A}_{m-1}\text{B}_m\text{O}_{3m+1})^2$, where A is a mono-, di-, or trivalent cation with dodecahedral coordination, B is a transition metal cation with octahedral coordination, and m is

the number of octahedral layers within the perovskite-like blocks [4]. This structure can accommodate a wide range of A- and B-site cation substitutions, allowing control of the physical properties.

Lead and bismuth-based Aurivillius compounds have drawn particular attention due to their favorable electrical properties and high ferroelectric Curie temperatures, with potential use in random access memory (RAM) [5]. The effect of the $6s^2$ lone pair of Pb^{2+} and Bi^{3+} induces cation shifts that break inversion symmetry, which can result in ferroelectricity in these compounds [6–8]. However, few Aurivillius phases exhibit both ferroelectric and ferromagnetic properties due to the contradictory character of the B-cation which is generally d^0 in ferroelectric compounds and d^n ($n \neq 0$) in magnetic compounds [9]. Therefore, studies with the aim of improving the properties of Aurivillius compounds have been focused in recent years especially on inducing the coexistence of both ferroelectric and ferromagnetic properties.

The introduction of first-row transition metal cations with a

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partially filled d -shell (d^n) on the perovskite B -site can potentially give rise to multiferroic properties in Aurivillius phases. It is well known that BiMnO_3 exhibits coupling between ferroelectricity and ferromagnetism [1,10,11]. Since it is believed that the ferroelectricity of perovskite-based compounds arises mainly from the distortion of BO_6 octahedra, the partial substitution of Mn^{3+} (d^4) for Nb^{5+} (d^0) on the Aurivillius B -site might improve the ferroelectric properties due to the effect of different ionic radii, as well as introducing a magnetic moment [12–14]. The substitution of Ln^{3+} cations on the Aurivillius A -site has previously been shown to improve the dielectric and ferroelectric properties [15,16].

Recently, we have reported on the novel series of compounds $\text{Pb}_{1-2x}\text{Bi}_{1.5+2x}\text{La}_{0.5}\text{Nb}_{2-x}\text{Mn}_x\text{O}_9$ synthesized by the molten salt method, for which single-phase products can be obtained for the compositions $0 \leq x \leq 0.3$ [17]. These compounds adopt a non-centrosymmetric crystal structure and cation disorder is found in both the bismuth oxide and perovskite blocks by Raman analysis, leading to greater structural distortion. The sample of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ with maximum Mn content exhibits the most distorted structure of the series and is expected to give the most pronounced magnetic properties.

Herein, we present a further structural investigation using neutron powder diffraction to study the atomic-scale disorder and the structural distortion that would favor enhanced ferroelectricity. The ferroelectric and magnetic properties of this compound are also investigated in detail to demonstrate the multiferroic properties.

2. Experimental procedure

Single-phase $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ was synthesized using a molten-salt method as reported in Ref. [17]. Neutron powder diffraction (NPD) data were collected on the High Resolution Powder Diffractometer (DN3) at the Section for Neutron Beam Technology of the Center for Science and Technology of Advanced Materials (BATAN), Indonesia. The wavelength of the incident neutron beam was 1.8195 Å and data were collected for 11 h at room temperature. The structural parameters were refined by the Rietveld technique using the RIETICA program [18]. For electrical measurements, the powder was pressed into pellets and sintered at 1173 K for 5 h. Silver conductive paste (Aldrich, 99%) was applied to both surfaces of the sintered pellet as the electrodes and heated at 423 K for 2 h. Capacitance was measured as a function of temperature using a precision LCR-meter (Agilent 4980A) with an amplitude of 1 V at different frequencies up to 1 MHz. Room temperature ferroelectric hysteresis was measured using a ferroelectric tester (TF analyzer, AixACCT) at 40 Hz. Magnetic susceptibility was measured using a SQUID magnetometer (Quantum Design MPMS XL7) in the temperature range from 5 to 300 K under applied magnetic fields of 1 T and 100 Oe. Magnetization as a function of applied field was measured from –5 T to 5 T at temperatures of 5 K and 150 K.

3. Results and discussion

$\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ has an orthorhombic structure with the $A2_1am$ space group as previously reported using XRD analysis [17]. Detailed structural parameters were obtained from Rietveld refinement of the NPD data. The initial structural model used was that of the parent compound $\text{PbBi}_2\text{Nb}_2\text{O}_9$ with the $A2_1am$ space group [19]. Here all atoms occupy general Wyckoff positions $8b$ except for $\text{Bi}(1)$ and $\text{O}(1)$ in the perovskite block which occupy $4a$ positions. It is well established that cations with lone pair electrons prefer to occupy the Bi-site in the bismuth oxide blocks, which suggests that Pb^{2+} may substitute in this block [8,20]. Disorder

between Mn and Nb is likely in the BO_6 octahedra. The perovskite A -site can be occupied by Pb, Bi and La.

Since the absence of an active lone pair of electrons limits the probability that La^{3+} occupies the Bi-site of the bismuth oxide blocks, which was also suggested by Raman spectroscopy as reported previously [17], we performed the initial refinement with a 50% occupancy of La^{3+} on the $\text{Pb}(1)$ site in the perovskite block. The site occupancies of $\text{Pb}(1)/\text{Bi}(1)$, $\text{Pb}(2)/\text{Bi}(2)$ and $\text{Nb}(1)/\text{Mn}(1)$ were then adjusted manually to optimize the fit while the atomic positions were refined automatically. Fig. 1 shows the Rietveld fits to the NPD data and the refined parameters are summarized in Table 1. The $\text{Bi}(2)$ sites in the bismuth oxide blocks are occupied by a random distribution of Pb^{2+} (12.5%) and Bi^{3+} (87.5%). Kennedy et al. [21] observed a similar preference of Pb^{2+} for the bismuth oxide blocks in $\text{Pb}_{1-x}\text{Sr}_x\text{Bi}_2\text{Nb}_2\text{O}_9$, even for low-Pb compositions. In the perovskite blocks of our current material, the $\text{Pb}(1)$ A -site is occupied by 50% La^{3+} , 35% Bi^{3+} , and 15% Pb^{2+} . This preference for a disordered distribution of Pb^{2+} and Bi^{3+} in both the bismuth oxide blocks and the perovskite A -site is also in agreement with earlier reports and is consistent with the similar electronic structures of the two cations [8,20]. According to the nominal formula of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$, the perovskite B -site should be occupied by Nb^{5+} and Mn^{3+} cations with fractions of 85% and 15%, respectively. Thus, the formula is perhaps more usefully written as $(\text{Bi}_{1.75}\text{Pb}_{0.25})\text{O}_2(\text{Pb}_{0.15}\text{Bi}_{0.35}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3})\text{O}_7$ in order to reflect the occupation of each cation in the bismuth and perovskite blocks. A representation of the crystal structure is shown in Fig. 2.

Since it is believed that the off-center distortion of the B -site cation is responsible for ferroelectricity in Aurivillius compounds, we focus on atomic displacements associated with the BO_6 octahedra. The $B(1)$ cation is displaced from the center of the octahedra along the c -axis resulting in longer $B(1)–\text{O}(1)$ and shorter $B(1)–\text{O}(2)$ bond distances as summarized in Table 2 (see also Fig. 2(c)). The B -site cations are also shifted along the a and b axes, evidenced by the different $B(1)–\text{O}(4)$ and $B(1)–\text{O}(5)$ distances. Displacement of the oxygen atoms from their ideal positions also results in tilting

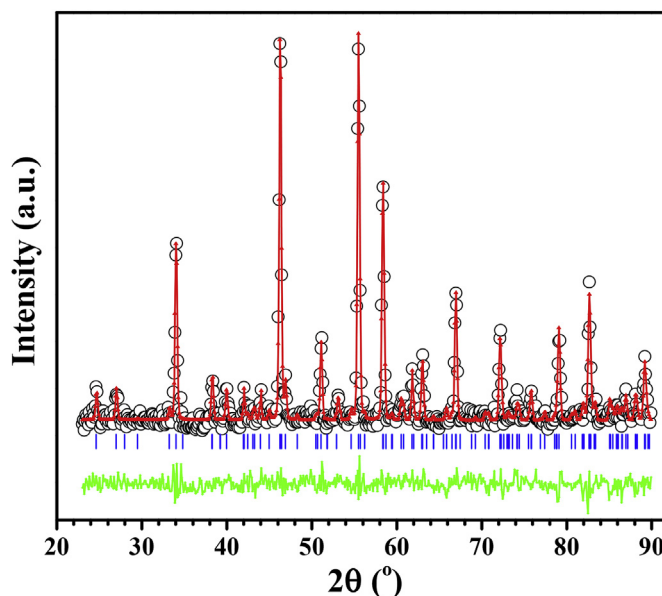


Fig. 1. Rietveld fits to NPD data for single-phase $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$: experimental data (black circles), calculated profile (red line), and difference profile (green line). The blue tick marks indicate the positions of allowed Bragg reflections in the space group $A2_1am$. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

Table 1

Refined structural parameters for $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ from NPD data at room temperature using the orthorhombic space group $A2_1am$, with $Z = 4$.

Atom	Site	x	y	z	Occ. (n)
Pb(1)	4a	0	0.2467(7)	0	0.15
Bi(1)	4a	0	0.2467(7)	0	0.35
La(1)	4a	0	0.2467(7)	0	0.5
Bi(2)	8b	0.4993(1)	0.7377(3)	0.2011(4)	0.875
Pb(2)	8b	0.4993(1)	0.7377(3)	0.2011(4)	0.125
Nb(1)	8b	0.4641(2)	0.7289(7)	0.4134(6)	0.850
Mn(1)	8b	0.4641(2)	0.7289(7)	0.4134(6)	0.150
O(1)	4a	0.4561(4)	0.1947(1)	0	1.0
O(2)	8b	0.4791(6)	0.7787(8)	0.3386(2)	1.0
O(3)	8b	0.7213(1)	0.0141(6)	0.25	1.0
O(4)	8b	0.7035(2)	0.9561(2)	0.0774(1)	1.0
O(5)	8b	0.7474(4)	0.9877(7)	0.5690(1)	1.0

$a = 5.4910(5)$ Å, $b = 5.4874(3)$ Å, $c = 25.1229(7)$ Å; $V = 757.0007(9)$ Å³.

$R_p = 14.04\%$, $R_{\text{bragg}} = 3.48$, $\chi^2 = 1.504$.

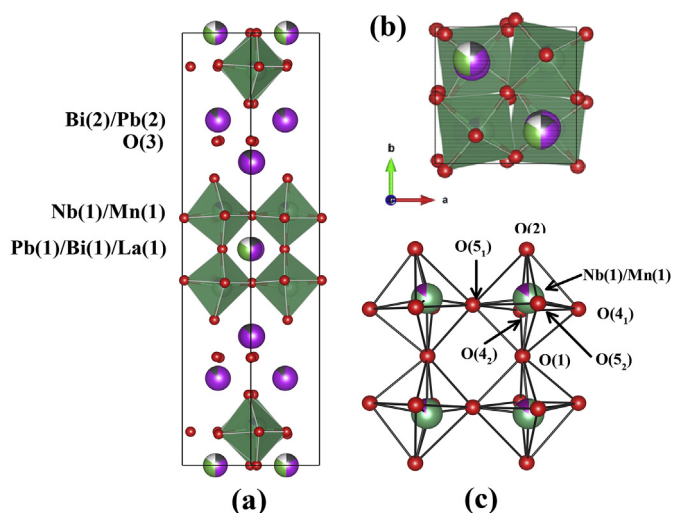


Fig. 2. (a) Orthorhombic structure of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ (ac -plane view). (b) Distortion of perovskite blocks projected in the ab plane. (c) View of linked BO_6 octahedra.

Table 2

Interatomic bond distances and bond angles associated with the BO_6 octahedra ($B = \text{Nb}$ or Mn).

Bond	Bond distance (Å)	Bond	Bond Angle (°)
$\text{B(1)}-\text{O(1)}$	2.1842(2)	$\text{O(2)}-\text{B(1)}-\text{O(1)}$	176.39
$\text{B(1)}-\text{O(2)}$	1.8996(2)	$\text{O(4)}-\text{B(1)}-\text{O(5)}$	159.42
$\text{B(1)}-\text{O(4)}$	1.7695(0)	$\text{B(1)}-\text{O(1)}-\text{B(1)}$	169.88
$\text{B(1)}-\text{O(4)}_2$	2.0056(4)	$\text{B(1)}-\text{O(5)}-\text{B(1)}$	153.46
$\text{B(1)}-\text{O(5)}_1$	2.0072(1)		
$\text{B(1)}-\text{O(5)}_2$	2.1525(2)		

of the BO_6 octahedra, as clearly seen in Fig. 2(b). This tilting leads to a more distorted crystal structure and results in a difference between the a and b lattice parameters. The degree of octahedral distortion can be quantified by the orthorhombicity ratio $(a-b)/(a+b)$ of 0.00033 [9]. This distortion is driven both by cation disorder and size mismatch between the perovskite A- and B-sites and further contributes to the ferroelectric displacement.

The morphology and size of the sample grains was previously investigated using SEM in Ref. [17]. The grains are highly anisotropic and plate-like, which is a typical feature of Aurivillius phases. The average grain size is in the range 2.7–3.1 μm .

The frequency dependence of the dielectric constant and

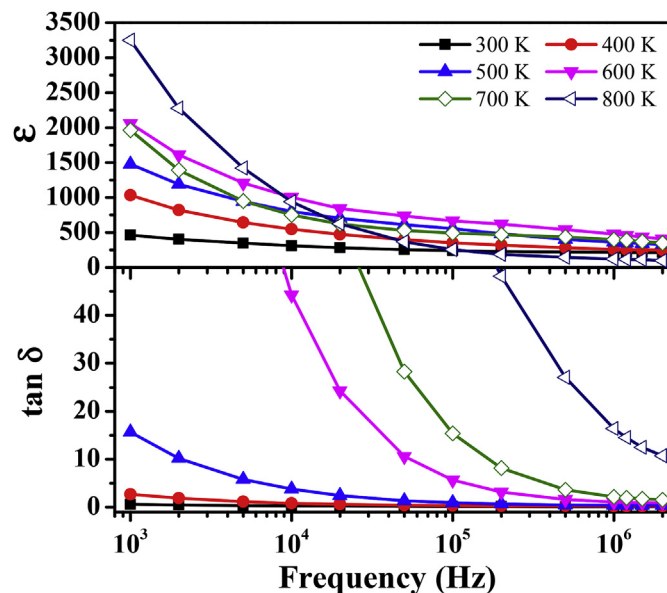


Fig. 3. Frequency dependence of dielectric constant (ϵ) and loss ($\tan \delta$) of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ at different temperatures.

dielectric loss of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ at different temperatures is shown in Fig. 3. It is observed that the dielectric constant decreases as frequency increases and becomes stable at frequencies above 10 kHz, which is consistent with the behavior of ordinary ferroelectric materials. The higher dielectric constant at lower frequencies is due to the contribution of different polarizations (i.e. dipolar, ionic, electronic, and interfacial) [22]. This behavior is also caused by the accumulation of charge carriers on the surface and at grain boundaries, explained by the Maxwell-Wagner effect [12]. Thus, high values of the dielectric constant at low frequencies are not only caused by intrinsic but also by extrinsic factors. It is also clearly seen in Fig. 3 that the dielectric constant initially increases with temperature and then significantly decreases from 600 K to 700 K, suggesting that the sample has a ferroelectric-paraelectric transition (T_c) in this temperature range. The dielectric loss increases slightly with temperature below 600 K, and then more rapidly at higher temperature, which can also indicate a transition to the paraelectric phase where the sample becomes more conducting. Moreover, at high temperature, the electrical conductivity is enhanced by thermal effects as is the case for most semiconductors, resulting in a significant increase of the dielectric loss [23].

Fig. 4 shows the temperature dependence of the dielectric constant (ϵ) and dielectric loss ($\tan \delta$) at high frequencies, which best reflect the intrinsic polarizability. A single peak in the dielectric constant is observed between 600 K and 640 K depending on frequency, corresponding to T_c . Compared with the parent compound $\text{PbBi}_2\text{Nb}_2\text{O}_9$ for which $T_c = 830$ K [24], the transition temperature is significantly decreased for $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$. Since the presence of perovskite A-site cations with $6s^2$ lone pairs such as Bi^{3+} and Pb^{2+} strongly favors a highly distorted structure and induces an increase in T_c [8,25], the substitution of La^{3+} for Pb^{2+} on the A-site reduces the concentration of A-site lone pair cations, leading to a smaller degree of distortion and hence a lower Curie temperature. The decrease of T_c is also consistent with the larger Goldschmidt tolerance factor (t) of the perovskite block of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ (0.940) compared to $\text{PbBi}_2\text{Nb}_2\text{O}_9$ (0.932) [6].

The dielectric peak exhibits a strong frequency dependence with

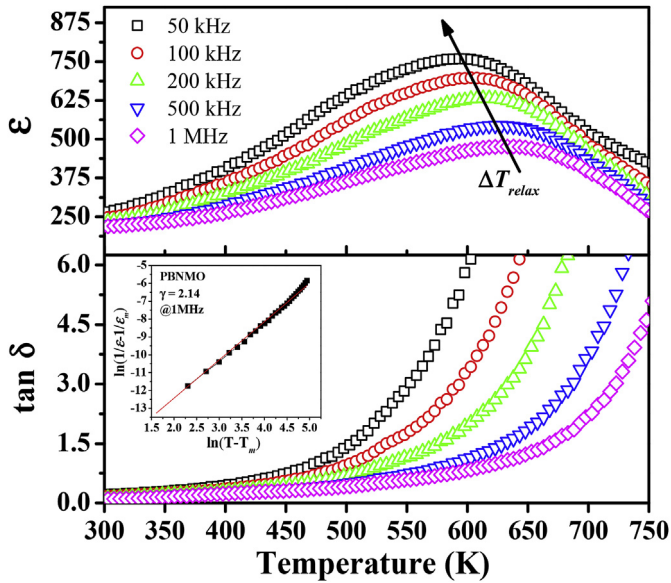


Fig. 4. Temperature dependence of dielectric constant (ϵ) and loss ($\tan \delta$) of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$. The inset shows a modified Curie-Weiss fitted line to quantify relaxor ferroelectric behavior.

a temperature relaxation $\Delta(T_{m(1\text{MHz})} - T_{m(50\text{kHz})})$ of around 40 K, which confirms pronounced relaxor-ferroelectric behavior [26]. This behavior can also be described by the degree of diffuseness (γ) using a modified Curie-Weiss law [15]. The inset of Fig. 4 shows a linear fit to a plot of $\ln(1/\epsilon_r - 1/\epsilon_m)$ versus $\ln(T - T_m)$ at 1 MHz, where the value of $\gamma = 1$ represents normal ferroelectric behavior and $\gamma = 2$ is typical for relaxor-ferroelectric. The fitted value of γ is 2.14. This relaxor behavior is strongly correlated with the compositional disorder on all the cation sites.

Room temperature P - E hysteresis loops measured at 40 Hz under various electric field ranges are shown in Fig. 5. At 60 kV/cm, the hysteresis loop is unsaturated, implying that higher electric fields are needed to align all the domains. At the highest applied

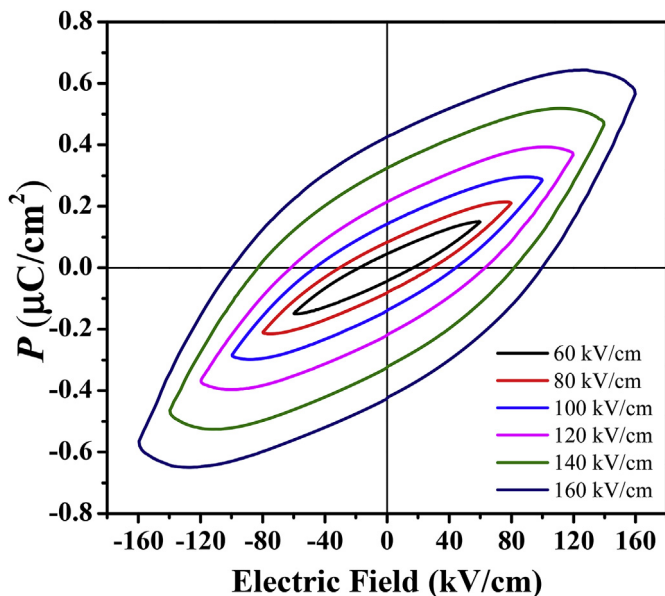


Fig. 5. P - E hysteresis loops of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ measured at room temperature and 40 Hz frequency.

electric field of 160 kV/cm before electrical breakdown occurred, the loop is still unsaturated, which is likely due to leakage currents [11,27]. The substitution of Mn^{3+} for Nb^{5+} in $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ increases the electron concentration and thus the electrical conductivity, hence the high leakage current and significant dielectric loss [23]. Although leaky, both the remnant polarization (P_r) and saturated polarization (P_s) increase with electric field in non-linear fashion. This behavior reflects domain wall motion associated with ferroelectricity [28]. The remnant polarization (P_r) and the coercive field (E_c) are $0.43 \mu\text{C}/\text{cm}^2$ and $99.7 \text{ kV}/\text{cm}$, respectively, under the maximum electric field of 160 kV/cm. The ferroelectric polarization in this sample is correlated with off-center displacement of the B -site cation in the a -direction, according to $A2_1am$ symmetry. It has previously been argued that octahedral tilting, as shown in Fig. 2(b), can further enhance the shift of this cation along the a -axis [9].

To investigate the magnetic properties of $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$, we performed magnetic susceptibility versus temperature and magnetization versus applied field measurements. Fig. 6(a) shows the magnetic susceptibility and inverse susceptibility in zero-field-cooled (ZFC) mode measured on warming in an applied magnetic field of 1 T. The ZFC susceptibility (χ) curve indicates paramagnetic-like behavior with no evidence of long-range magnetic ordering, as evidenced by the monotonous decrease of χ with increasing temperature. However, a Curie-Weiss fit to the linear region of the $1/\chi$ - T curve above 100 K yields a positive Curie-Weiss temperature (θ_{CW}) of 33.44 K, indicating the predominance of ferromagnetic interactions. To verify the existence of ferromagnetic interactions, the ZFC susceptibility was measured in a lower applied field of 100 Oe as shown in Fig. 6(b). An obvious ferromagnetic-paramagnetic transition at $T_c = 33.5 \text{ K}$ is now observed, where T_c is obtained from the maximum of the $(d\chi/dT)$ curve; this temperature is consistent with the fitted θ_{CW} . This is also close to the value of T_c (21.5 K) for the Mn-doped Aurivillius phase $\text{PbBi}_4\text{Ti}_4\text{O}_{15}$, where 15% of Ti^{4+} on the B -site was replaced by Mn [12].

The effective moment (μ_{eff}) per Mn ion calculated from the Curie-Weiss fit is $4.38 \mu_B$. The spin-only values of μ_{eff} for high-spin Mn^{3+} and Mn^{4+} are $4.90 \mu_B$ and $3.87 \mu_B$, respectively [29,30]. Thus, the sample likely contains approximately equal proportions of Mn^{3+} and Mn^{4+} . The mixed-valent $\text{Mn}^{3+}/\text{Mn}^{4+}$ may enable double-exchange to take place, giving rise to the weakly ferromagnetic behavior. The magnetic ordering will be short-range in nature because only 15% of the B -site is occupied by Mn, far below the percolation threshold. Furthermore, strong ferromagnetic behavior has mainly been observed in Aurivillius phases with a thicker perovskite block than our two-layer $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$, such as the four-layered structure, and can be enhanced by combining two magnetic cations in a 1:1 ratio [31,32]. We speculate that the double exchange pathway may be in the ab -plane via $\text{Mn}(1)\text{-O}(4)\text{-Mn}(1)$ and $\text{Mn}(1)\text{-O}(5)\text{-Mn}(1)$, which are depicted in Fig. 2(c). These exchange interactions might occur in locally Mn-rich areas, the presence of which was previously suggested by the vibrational mode at 756 cm^{-1} in Raman spectra [17]. A preference for local ordering of the magnetic cation was also observed previously in $\text{SrBi}_2\text{Nb}_{2-x}\text{Fe}_x\text{O}_9$, where (Fe-O-Fe) super-exchange interactions are of relevance [33]. Such local ordering is also evidenced by the high dielectric loss and leakage current in $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ [23], which are attributed to double exchange involving $\text{Mn}^{3+}\text{-O-Mn}^{4+}$.

Fig. 7 shows magnetization versus applied field $M(H)$ measurements for $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ at 5 K and 150 K. The magnetization does not reach saturation in the applied field range -5 T to 5 T . The $M(H)$ curve at 5 K adopts the typical S-shaped, narrow hysteresis loop of a weak ferromagnetic, where M_r and H_c

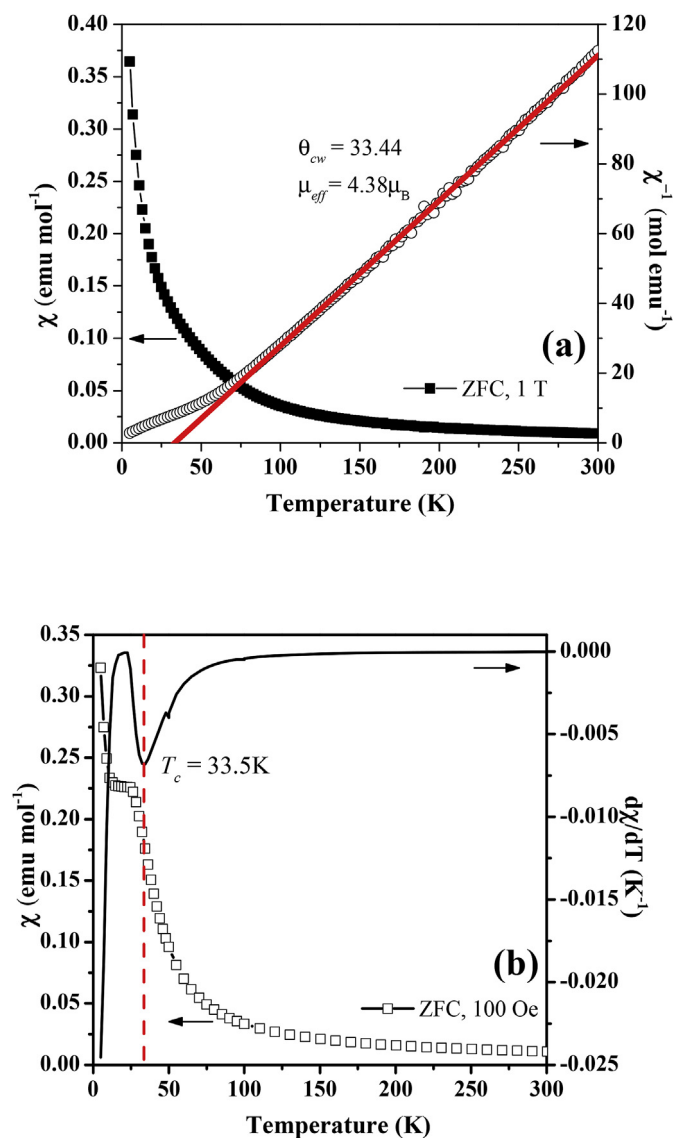


Fig. 6. (a) Temperature dependence of ZFC magnetic susceptibility (χ) and inverse magnetic susceptibility ($1/\chi$) measured in an applied field of 1 T. The red line is a linear fit to the experimental data using the Curie-Weiss law. (b) ZFC magnetic susceptibility (χ) and its derivative ($d\chi/dT$) measured in an applied field of 100 Oe. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

are estimated to be 0.01 emu/g and 67 Oe. At 150 K, where the sample is far above the T_c of 33 K, linear behavior is observed with no loop opening, as typical for paramagnetic behavior.

4. Conclusion

A single-phase sample of the Aurivillius material $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ has been synthesized by a molten salt method using $\text{K}_2\text{SO}_4/\text{Na}_2\text{SO}_4$. Rietveld refinement using neutron diffraction data suggested considerable cation disorder, where La atoms partially occupy the perovskite A-site, Pb/Bi atoms occupy both the perovskite A-site and bismuth oxide block, and Nb/Mn occupy the perovskite B-site. The dielectric constant exhibits a broad frequency-dependent peak indicating a ferroelectric transition temperature (T_c) of 600–640 K, which is typical for a relaxor ferroelectric. The disorder associated with the simultaneous

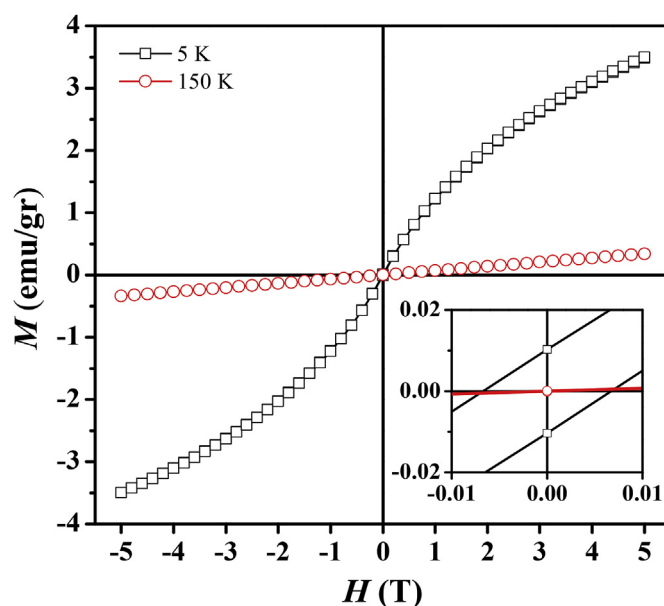


Fig. 7. Magnetization versus applied field for $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ measured at 5 K and 150 K.

substitution of La^{3+} and Mn^{3+} in $\text{PbBi}_2\text{Nb}_2\text{O}_9$ is likely responsible for the relaxor ferroelectric behavior. Ferroelectric P - E loops remain unsaturated in applied electric fields up to 160 kV/cm and a maximum P_r of 0.43 $\mu\text{C}/\text{cm}^2$ was obtained. Magnetic susceptibility measurements indicated that manganese is present in the mixed-valence state of $\text{Mn}^{3+}/\text{Mn}^{4+}$. A positive Curie-Weiss temperature and a peak in the susceptibility at 33 K indicate the presence of ferromagnetic interactions and short-range order which are assumed to arise from double-exchange involving Mn^{3+} -O- Mn^{4+} bonds in the BO_6 octahedra. In conclusion, the Aurivillius compound $\text{Pb}_{0.4}\text{Bi}_{2.1}\text{La}_{0.5}\text{Nb}_{1.7}\text{Mn}_{0.3}\text{O}_9$ exhibits the coexistence of both ferroelectric and ferromagnetic properties facilitating applications in memory storage devices.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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